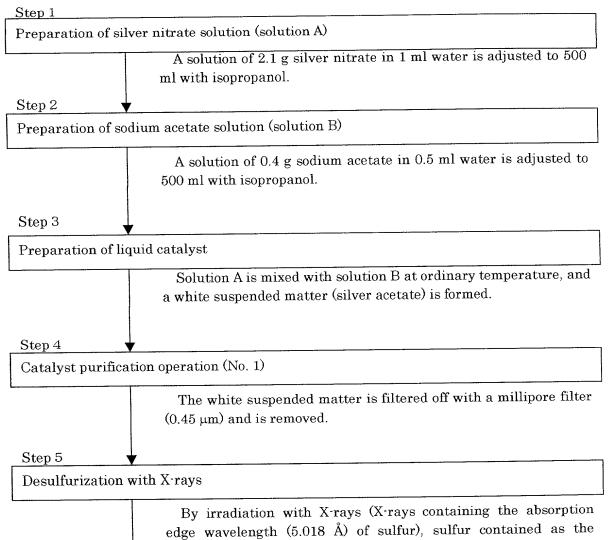
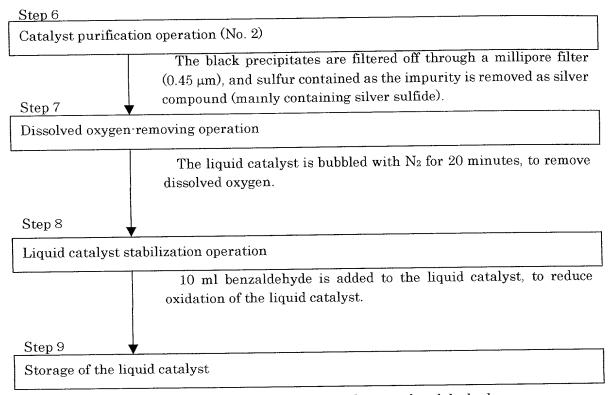
Fig.1A



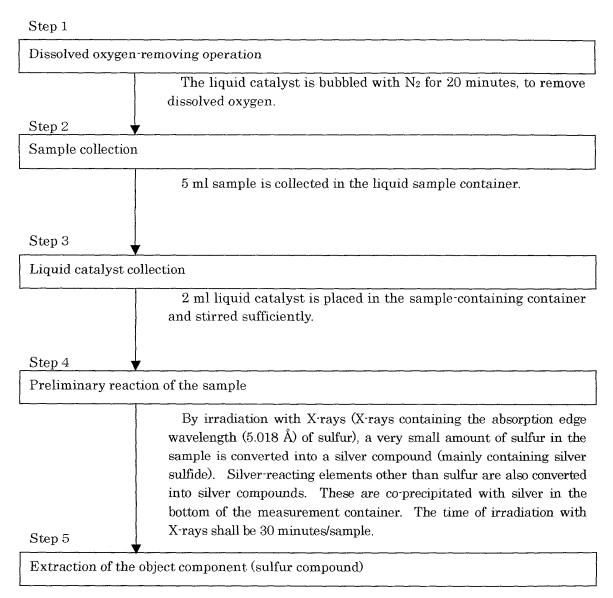
By irradiation with X-rays (X-rays containing the absorption edge wavelength (5.018 Å) of sulfur), sulfur contained as the impurity is changed into a silver compound containing silver sulfide. The time of irradiation with X-rays is 1 hour (for 15 ml container). The sample is then left.

Fig.1B



The liquid catalyst is stored in a cool and dark place.

Fig.2



In the presence of co-sedimented silver compounds and silver, scattered X-rays are increased and the absorption of a characteristic X-ray of S occurs, thus making analysis of a very small amount of sulfur difficult. Accordingly, ammonia or aldehyde is added to the sample and then left at a constant temperature (30 °C) for about 20 hours. By this operation, silver compounds and silver other than the object component are dissolved, diffused and removed from the sample measurement surface.

Fig. 3

Fig. 4

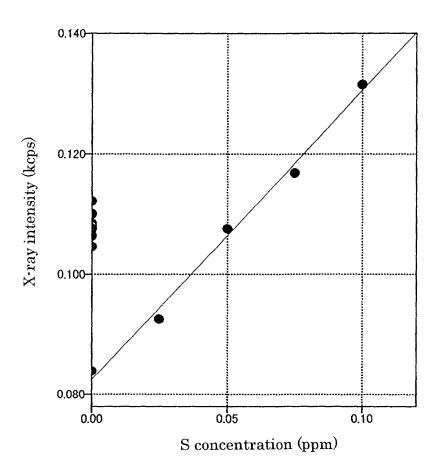
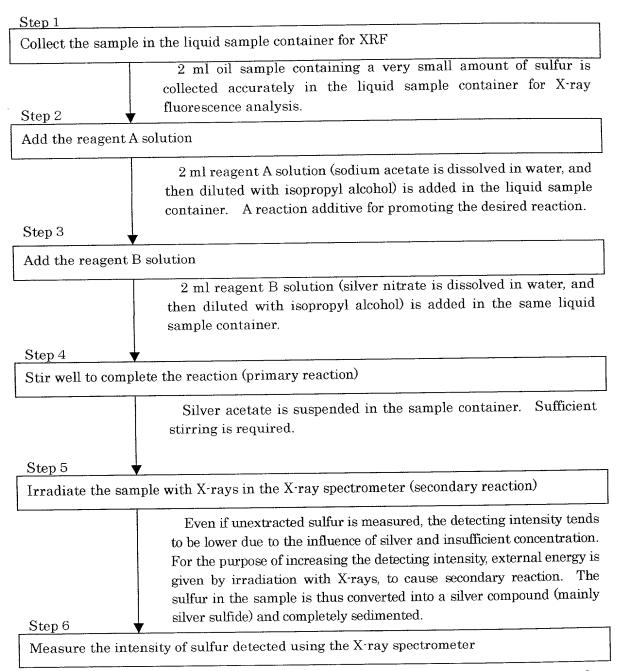


Fig.5



By upward irradiation to precipitates, the conc. solution can be measured to make analysis at 30 ppb level feasible.

Fig. 6

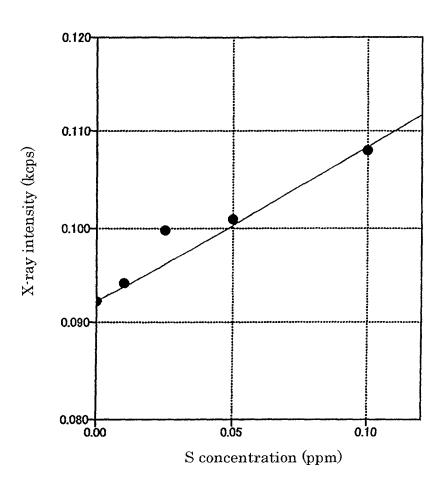


Fig. 7

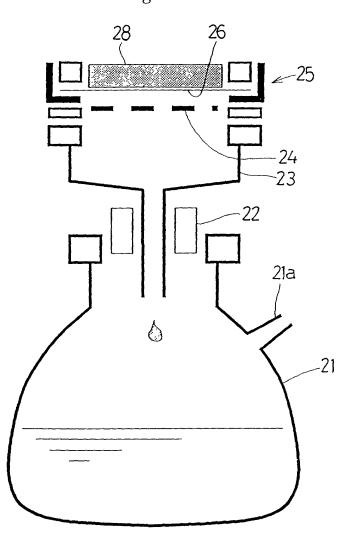


Fig. 8

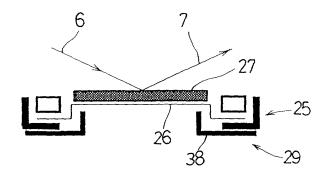


Fig. 9

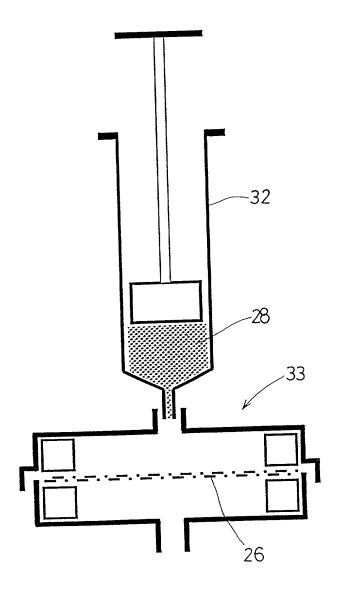


Fig. 10

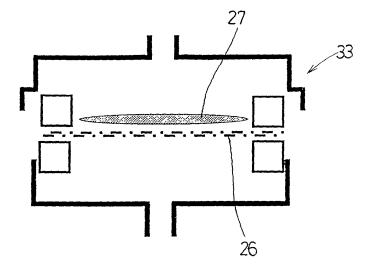


Fig. 11

